

june 2014 exam grd 11 physical science File PDF

June 2014 Exam Grade 11 Physical Science

Preparing for the June 2014 Grade 11 Physical Science exam requires a comprehensive understanding of the core concepts, exam structure, and effective revision strategies. Whether you're a student aiming to excel or a teacher seeking to guide learners, this article provides an in-depth overview of the key topics, exam tips, and resources to help you succeed in this important assessment.

Overview of the June 2014 Grade 11 Physical Science Exam

Understanding the scope and format of the June 2014 Physical Science exam is essential for effective preparation. The exam typically assesses students' knowledge across various fundamental topics, emphasizing both theoretical understanding and practical application.

Exam Structure and Duration

- Total Duration: 3 hours
- Number of Questions: Varies, generally including multiple-choice, short-answer, and essay-style questions
- Sections:
 - Physics (Mechanics, Waves, Electricity)
 - Chemistry (Atomic Structure, Chemical Reactions, Periodic Table)
 - Practical/Experimental Questions

Marking Scheme

- Multiple-choice questions: 1 mark each
- Short-answer questions: 2-4 marks each
- Extended response questions: up to 10 marks
- Practical-based questions: assess application and analysis skills

Key Topics Covered in the June 2014 Exam

A thorough review of the core topics will help you identify areas to focus on. Here are the main sections typically covered:

Physics Topics

- Mechanics
 - Motion and graphs of motion
 - Newton's Laws of Motion
 - Force, mass, and acceleration
 - Work, energy, and power
 - Momentum and collisions
- Waves and Sound
 - Wave properties (wavelength, frequency, amplitude)
 - Types of waves (transverse, longitudinal)
 - Sound propagation and Doppler effect
- Electricity and Magnetism
 - Electric circuits (series and parallel)
 - Ohm's Law and resistance
 - Electromagnetism basics

Chemistry Topics

- Atomic Structure
 - Protons, neutrons, electrons
 - Isotopes and atomic mass
 - Electron configurations
- Chemical Bonding and Reactions
 - Ionic, covalent, and metallic bonds
 - Types of chemical reactions (synthesis, decomposition, displacement)
 - Balancing chemical equations

- Periodic Table
- Groups and periods
- Properties of metals and non-metals
- Trends in reactivity and atomic radius

Practical Skills and Experimental Knowledge

- Interpreting graphs and data
- Conducting basic experiments safely
- Understanding scientific methods and reporting results

Effective Study Tips for the June 2014 Grade 11 Physical Science Exam

Maximizing your revision time and employing strategic study techniques can significantly improve your exam performance.

1. Create a Study Plan

- Break down topics into manageable sections
- Allocate specific days for each topic
- Include revision and practice tests in your schedule

2. Use Past Exam Papers and Practice Questions

- Familiarize yourself with the exam format
- Practice under timed conditions
- Review answers and understand mistakes

3. Focus on Understanding Concepts

- Avoid rote memorization; aim to grasp the underlying principles
- Use diagrams and visual aids to clarify complex ideas
- Relate concepts to real-world applications

4. Master Key Formulas and Definitions

- Physics formulas: velocity, acceleration, work, power, etc.
- Chemistry definitions: molar mass, atomic number, valency
- Keep a formula sheet for quick revision

5. Engage in Group Discussions and Tutorials

- Explaining concepts to peers reinforces your understanding
- Clarify doubts with teachers or online resources

Additional Resources for June 2014 Physical Science Revision

Utilize the following materials to enhance your preparation:

- **Textbooks:** Ensure you have the prescribed textbooks from your syllabus.
- **Online Platforms:** Websites like Khan Academy, Coursera, and educational YouTube channels.
- **Study Guides:** Focused revision notes and summaries tailored for Grade 11 Physical Science.
- **Past Exam Papers:** Available through your school or educational websites to practice exam-style questions.

Common Challenges and How to Overcome Them

Understanding typical difficulties students face can help you prepare more effectively:

Difficulty in Applying Concepts

- Practice a variety of questions that require application
- Use diagrams and real-life examples to contextualize theory

Time Management During the Exam

- Practice timed tests
- Prioritize questions based on marks allocated

Understanding Practical Components

- Conduct experiments if possible
- Review lab reports and practical exam guidelines

Summary and Final Tips

Preparing for the June 2014 Grade 11 Physical Science exam demands dedication, strategic planning, and thorough understanding. Remember:

- Review all core topics systematically
- Practice past papers under exam conditions
- Focus on understanding, not just memorization
- Clarify doubts promptly
- Stay calm and confident during the exam

Achieving success in your Physical Science exam is within your reach with consistent effort and effective strategies. Use this guide as a roadmap to steer your revision process, and aim for excellence on exam day!

June 2014 Exam Grade 11 Physical Science: A Comprehensive Guide and Analysis

Preparing for the June 2014 Exam Grade 11 Physical Science can be a challenging yet rewarding experience for students aiming to excel in their academic journey. With a focus on key concepts, exam strategies, and common pitfalls, this guide aims to equip learners with a detailed understanding of what to expect, how to prepare effectively, and how to approach the exam confidently. Whether you're revisiting foundational principles or tackling complex questions, this detailed analysis will help you navigate the exam with clarity and purpose.

Introduction to the June 2014 Exam Grade 11 Physical Science

The June 2014 Exam Grade 11 Physical Science was designed to assess students' understanding of core topics such as matter, energy, atomic structure, chemical reactions, and physical principles governing motion and forces. The exam typically tests both theoretical knowledge and practical application skills, emphasizing problem-solving, calculations, and conceptual clarity.

Understanding the structure of the exam and the types of questions asked can significantly improve your preparation strategy. This guide will delve into the key areas covered, common question formats, and tips for maximizing your performance.

Overview of the Exam Structure

Sections of the Exam

The exam generally comprises multiple sections, each focusing on different aspects of physical science:

- Multiple Choice Questions (MCQs): Test foundational knowledge and quick reasoning.
- Short Answer Questions: Require concise explanations, calculations, or definitions.
- Extended Response or Problem-solving: Involves more detailed calculations, explanations, or practical applications.

Duration and Marking Scheme

- Total Duration: Usually 2.5 to 3 hours.
- Total Marks: Typically around 100 marks, distributed across sections.
- Time Management: Allocate time proportionally to sections; for example, spend more time on problem-solving questions.

Key Topics Covered in the June 2014 Exam

- Matter and Its Properties

Understanding states of matter, atomic structure, and bonding

- Atomic models: Dalton, Thomson, Rutherford, Bohr.
- Electron configuration and atomic number.
- Types of bonding: ionic, covalent, metallic.
- States of matter: solids, liquids, gases, and plasma.
- Changes of state and phase diagrams.

- Chemical Reactions and Equations

Balancing equations, types of reactions, and factors affecting reactions

- Types: synthesis, decomposition, single replacement, double replacement, combustion.
- Conservation of mass.
- Reaction rates and factors influencing them.
- Acids, bases, and pH concepts.
- Oxidation and reduction.

- Energy and Work

Forms of energy, law of conservation, and practical applications

- Kinetic and potential energy.
- Energy transformations.
- Work done and power.
- Energy efficiency.

- Forces and Motion

Newton's laws, equations of motion, and graphs

- Types of forces: gravitational, frictional, normal, tension.
- Newton's laws of motion.
- Equations of uniformly accelerated motion.
- Free-body diagrams.
- Momentum and impulse.

- Electricity and Magnetism

Circuit concepts, Ohm's law, magnetic fields

- Series and parallel circuits.
- Resistance, voltage, current calculations.
- Electromagnetism basics.
- Magnetic forces on currents.

- Waves and Optics

Wave properties, reflection, refraction, and lenses

- Types of waves: transverse and longitudinal.
- Wave speed, frequency, wavelength.
- Reflection and refraction principles.
- Lens formula and magnification.

Common Question Formats and How to Approach Them

Multiple Choice Questions (MCQs)

- Focus on core facts and quick reasoning.
- Read all options carefully before selecting.
- Use process of elimination to narrow down choices.

Short Answer Questions

- Be concise but complete.
- Show all calculations clearly.
- Use appropriate units and symbols.

Extended Response / Problem-Solving

- Read the question carefully.
- Identify what is asked (concept, calculation, explanation).
- Plan your solution: outline steps before solving.
- Label diagrams where applicable.
- Check units and significant figures in calculations.
- Review your answer for accuracy and completeness.

Strategies for Effective Preparation

Understand Key Concepts

- Focus on understanding rather than memorization.
- Use diagrams, charts, and models to visualize concepts.
- Practice explaining concepts in your own words.

Practice Past Exam Papers

- Review previous exams, including the June 2014 paper.
- Time yourself to improve speed.
- Identify recurring question types and topics.

Revise Core Calculations

- Master formulas related to motion, energy, electricity, and reactions.
- Practice common calculations: molar calculations, energy problems, circuit analysis.

Clarify Difficult Topics

- Seek help from teachers or study groups for challenging areas.
- Use online resources and tutorials for additional explanations.

Develop Exam Strategies

- Read questions carefully.
- Allocate time proportionally.
- Leave difficult questions for last.
- Review answers if time permits.

Common Pitfalls and How to Avoid Them

- Misreading questions: Always read carefully; underline key parts.
- Ignoring units: Units are crucial for correctness.
- Incorrect calculations: Double-check arithmetic and formulas.
- Incomplete answers: Answer all parts of multi-part questions.
- Time mismanagement: Practice pacing during revisions.

Sample Questions and Model Answers

Question 1: Atomic Structure

Describe the main differences between Rutherford's and Bohr's atomic models.

Model Answer:

Rutherford's atomic model proposed that atoms consist of a tiny, dense nucleus surrounded by electrons, explaining the scattering of alpha particles. However, it did not specify electron arrangements. Bohr's model built upon Rutherford's by introducing quantized energy levels for electrons orbiting the nucleus, which explained atomic spectra. The key difference is that Bohr's model incorporates specific electron shells, whereas Rutherford's model does not.

Question 2: Force Calculation

A 5 kg object accelerates at 3 m/s². Calculate the force acting on it.

Model Answer:

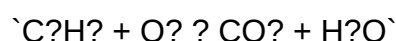
Using Newton's second law:

Force (F) = mass (m) × acceleration (a)

$$F = 5 \text{ kg} \times 3 \text{ m/s}^2 = 15 \text{ N}$$

Question 3: Chemical Equation Balancing

Balance the following chemical reaction:

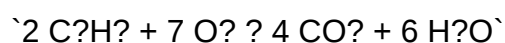


Model Answer:

Balance carbons: 2 C on the left, so 2 CO₂ on the right.

Balance hydrogens: 6 H on the left, so 3 H₂O on the right.

Balance oxygens: On the right, 2×2 (from CO₂) + 3 (from H₂O) = 7 O atoms. On the left, O₂ molecules: 7 O atoms, so 7/2 O₂ molecules. To avoid fractions, multiply entire equation by 2:



Final Tips for Success in the June 2014 Exam Grade 11 Physical Science

- Start your revision early to cover all topics thoroughly.
- Use a variety of resources: textbooks, past papers, online tutorials.
- Focus on understanding concepts, not just memorizing facts.
- Practice under exam conditions to build confidence.
- Stay calm and positive during the exam.

Conclusion

The June 2014 Exam Grade 11 Physical Science offers an opportunity to demonstrate your understanding of fundamental scientific principles. By thoroughly reviewing key topics, practicing past papers, and developing effective exam strategies, you can approach the exam with confidence and aim for excellent results. Remember, consistent effort and a clear understanding are the keys to success in physical science examinations. Good luck!

QuestionAnswer What were the main topics covered in the June 2014 Grade 11 Physical Science exam? The exam primarily focused on topics such as atomic structure, chemical bonding, thermodynamics, electrochemistry, and basic principles of mechanics relevant to physical science. How can students effectively prepare for the June 2014 Grade 11 Physical Science exam? Students should review their class notes, practice past exam papers, understand key concepts and formulas, and engage in group discussions to reinforce understanding of core topics. What are common challenges students faced in the June 2014 Physical Science exam? Common challenges included applying theoretical concepts to practical problems, balancing calculations and explanations, and managing time during the exam. Are there specific topics in the June 2014 exam that students should focus on for better scores? Yes, students should pay particular attention to electrochemical cells, thermodynamics calculations, and atomic structure questions, as these areas often carry significant marks. What resources are recommended for revising the June 2014 Grade 11 Physical Science exam content? Recommended resources include past exam papers, textbooks aligned with the syllabus, online tutorials, and revision guides specific to Grade 11 Physical Science. How did the June 2014 exam differ from previous years for Grade 11 Physical Science? The June 2014 exam included more application-based questions and integrated concepts from different chapters, requiring students to demonstrate both understanding and analytical skills. What strategies can students use to improve their performance in Physical Science exams like June 2014? Students should practice time management, focus on understanding concepts rather than rote memorization, and regularly test themselves with past papers to identify weak areas.

Related keywords: June 2014 exam, Grade 11 Physical Science, exam paper, past exam questions, syllabus, study guide, physical science topics, exam tips, marking scheme, revision notes

Physical Science Creative Teaching Press Physical Changes

Physical science creative teaching press physical changes is a vital topic in the realm of science education, especially for educators aiming to foster understanding and curiosity about the fundamental nature of matter. Understanding physical changes is essential for students to grasp how substances transform without altering their chemical composition. This article explores the concept of physical changes in physical science, emphasizing creative teaching methods, key principles, real-world examples, and effective strategies to make learning engaging and comprehensive.

Understanding Physical Changes in Physical Science

What Are Physical Changes?

Physical changes refer to alterations in the form, appearance, or state of a substance that do not modify its chemical identity. These changes are often reversible, meaning the original substance can typically be recovered without chemical reactions taking place.

Examples of physical changes include:

- Melting ice into water
- Boiling water into vapor
- Cutting paper
- Dissolving sugar in tea
- Crushing a can

Key characteristics of physical changes:

- No new substances formed
- Usually reversible
- Involve changes in physical properties like shape, size, or phase

Difference Between Physical and Chemical Changes

While physical changes involve no change in chemical composition, chemical changes result in the formation of new substances with different properties.

Aspect	Physical Change	Chemical Change
--------	-----------------	-----------------

|-----|-----|-----|

| Definition | Change in physical form or state | Formation of new substances |

| Reversibility | Usually reversible | Usually irreversible |

| Example | Melting ice | Burning wood |

Creative Teaching Strategies for Physical Changes

Engaging students in understanding physical changes requires innovative and creative teaching methods. Here are some effective strategies:

Hands-On Experiments and Demonstrations

Practical activities help students visualize and understand physical changes concretely.

Examples include:

- Melting ice cubes and observing the change
- Evaporating saltwater
- Crumpling aluminum foil
- Freezing water to form ice
- Dissolving sugar and salt

Tips for effective experiments:

- Use clear, simple materials
- Encourage students to record observations
- Discuss reversible vs. irreversible changes

Interactive Visual Aids and Simulations

Utilize multimedia tools like animations and virtual labs to demonstrate physical changes that are difficult to observe directly.

Benefits include:

- Enhancing visualization skills
- Providing safe environments for experimentation
- Catering to diverse learning styles

Storytelling and Real-Life Contexts

Integrate stories or real-world examples to make the concept relatable.

Examples:

- How ice melts outside on a sunny day
- How a puddle evaporates after rain
- The process of freezing water into ice cubes at home

Creative Classroom Activities

Engage students through activities that promote exploration and discussion.

Ideas include:

- "Change Detective" game where students identify physical changes around them
- Group projects on different physical changes observed in nature
- Art projects like sculpting or paper folding to demonstrate physical transformations

Key Concepts and Principles of Physical Changes

States of Matter and Phase Changes

Understanding how matter transitions between solid, liquid, and gas states is central to physical science.

Common phase changes:

- Melting (solid to liquid)
- Freezing (liquid to solid)
- Vaporization or boiling (liquid to gas)
- Condensation (gas to liquid)
- Sublimation (solid to gas)

Teaching tip: Use diagrams and models to illustrate these transitions dynamically.

Properties of Materials During Physical Changes

Monitor how physical properties such as shape, volume, density, and color change during physical transformations.

Examples:

- Volume expansion of metal when heated
- Changes in shape during molding or crushing
- Color changes due to temperature effects

Reversibility and Conservation of Mass

Most physical changes are reversible and adhere to the law of conservation of mass, meaning the total mass remains constant during the change.

Discussion points:

- How physical changes do not alter the chemical identity
- Examples demonstrating conservation of mass in physical changes

Real-World Applications of Physical Changes

Understanding physical changes has numerous applications across industries and daily life.

In Manufacturing and Industry

- Metal forging and shaping
- Plastic molding
- Food processing (freezing, boiling, drying)

In Nature and Environment

- Water cycle processes like evaporation and condensation
- Glacial melting

- Weathering of rocks

In Daily Life

- Cooking (boiling, freezing)
- Cleaning (dissolving detergents)
- Recycling materials like aluminum cans

Assessment and Evaluation Strategies

To gauge understanding, educators can implement various assessment tools:

- Quizzes and Worksheets: Focused on identifying physical changes
- Observation and Lab Reports: Documenting experiments
- Group Discussions: Explaining processes and concepts
- Creative Projects: Designing models or presentations illustrating physical changes

Challenges and Solutions in Teaching Physical Changes

Common challenges include:

- Abstract concepts difficult for students to grasp
- Confusing physical changes with chemical changes
- Limited access to laboratory resources

Solutions:

- Use visual aids and models extensively
- Reinforce concepts with real-life examples
- Incorporate virtual labs and simulations
- Encourage critical thinking and questioning

Conclusion

Teaching physical science with a focus on physical changes demands creativity, engagement, and clarity. By employing hands-on experiments, visual tools, storytelling, and real-world contexts, educators can foster a deep understanding of physical changes among students. Recognizing the

importance of these concepts not only enhances scientific literacy but also prepares learners to appreciate the dynamic nature of matter in everyday life and various industries. As educators continue to innovate in their teaching approaches, the topic of physical changes remains a fundamental and exciting area of physical science education.

Keywords for SEO Optimization:

- Physical science
- Creative teaching methods
- Physical changes
- Science education
- Phase changes
- Reversible changes
- Hands-on science experiments
- Teaching physical science effectively
- Real-world science applications

Physical Science Creative Teaching Press Physical Changes: A Comprehensive Guide to Engaging Science Education

In the realm of physical science education, understanding the concept of physical changes is fundamental. When educators utilize resources from the Creative Teaching Press, particularly their materials on physical science, they open doors to innovative, hands-on learning experiences. This guide delves into the essence of physical changes, exploring how teachers can effectively introduce, demonstrate, and assess this critical scientific concept using creative teaching strategies and resources.

Understanding Physical Changes in Science

Physical changes refer to alterations in the state or appearance of a substance without changing its chemical composition. These changes are often reversible and are fundamental to students' grasp of material properties, states of matter, and the nature of physical phenomena.

Key Characteristics of Physical Changes:

- No new substance forms
- Usually reversible
- Changes involve physical properties like shape, size, phase, or texture
- Do not affect the chemical identity of the substance

Examples include melting, freezing, boiling, condensation, breaking, and dissolving.

Why Focus on Physical Changes?

Teaching about physical changes lays the groundwork for understanding more complex topics like chemical reactions and molecular structure. It also encourages observational skills, critical thinking, and science process skills such as hypothesizing, experimenting, and analyzing.

Using creative teaching methods, including hands-on activities, visual aids, and real-world applications, enhances student engagement and comprehension.

Effective Strategies for Teaching Physical Changes

- Hands-On Experiments and Demonstrations

Experiments are the cornerstone of science teaching. Engaging students with tangible experiences helps solidify abstract concepts.

Sample Activities:

- Melting Ice: Show how ice melts into water when heated, emphasizing phase change from solid to liquid.

- Dissolving Salt or Sugar: Demonstrate solubility and how substances can undergo physical changes without chemical reactions.

- Crushing or Breaking: Break chalk or plastic to illustrate physical alteration, emphasizing that the material remains the same.

Tips:

- Use clear, simple materials.
- Encourage students to record observations and hypothesize outcomes.
- Discuss reversibility after each activity.

- Visual Aids and Graphic Organizers

Visual tools help students visualize the concept and organize their understanding.

Examples:

- Charts comparing physical and chemical changes
 - Diagrams illustrating phase changes
 - Flowcharts depicting the steps in various physical processes
-
- Incorporate Creative Media and Resources

Creative Teaching Press offers a suite of resources, including posters, activity books, and digital content, designed to make learning about physical changes interactive and memorable.

Utilize:

- Illustrated posters showing different physical changes
 - Interactive worksheets that challenge students to classify changes
 - Story-based scenarios that involve physical change concepts
-
- Real-Life Applications and Contextual Learning

Connecting science to students' everyday experiences makes lessons relevant and meaningful.

Examples:

- Melting ice in a beverage
- Freezing water to make ice cubes
- Dissolving sugar in coffee or tea
- Physical alterations in manufacturing or cooking

Classroom Activities and Lesson Plans

Here's a sample outline for a comprehensive lesson:

Lesson Title: Exploring Physical Changes

Objectives:

- Define physical change
- Identify different types of physical changes
- Conduct simple experiments to observe physical changes

Materials Needed:

- Ice cubes
- Water
- Salt
- Sugar
- Plastic or glass containers
- Crayons or chalk
- Safety goggles

Procedure:

- Introduction (10 minutes): Discuss what physical changes are, using visuals.
- Demonstration (15 minutes): Melting ice and dissolving salt/sugar.
- Student Activity (30 minutes): Students perform experiments:
 - Observe ice melting
 - Dissolve sugar in water
 - Break chalk into pieces
 - Freeze water and observe ice formation
- Discussion (15 minutes): Students share observations, compare results, and classify each change.
- Assessment (10 minutes): Quiz or worksheet to reinforce understanding.

Assessment Tips:

- Use questioning strategies to assess understanding.
- Have students draw diagrams of their experiments and label physical changes.

Assessing Understanding of Physical Changes

Assessment should gauge both conceptual understanding and practical skills.

Methods:

- Observation during activities
- Student reflections and explanations
- Quizzes with multiple-choice and short-answer questions
- Concept maps illustrating types of changes

Common Misconceptions and How to Address Them

- Misconception: Physical changes always involve a change in the substance's chemical identity.

Correction: Emphasize that physical changes do not alter the chemical structure, using examples like melting ice or tearing paper.

- Misconception: All physical changes are reversible.

Correction: Clarify that some physical changes, like breaking or cutting, are reversible, but others, like tearing or dissolving, may not be easily reversible.

Integrating Creative Teaching Press Resources

Creative Teaching Press provides various materials tailored for physical science instruction, which can elevate your teaching approach:

- Activity Books: Ready-made experiments and activities that align with standards.
- Posters and Visuals: Engaging images that reinforce concepts.
- Interactive Worksheets: Opportunities for students to classify and analyze physical changes.
- Digital Resources: Online games and quizzes to supplement lessons.

Strategies for Integration:

- Use posters as classroom decor to continually reinforce concepts.
- Incorporate activity books into small-group work or centers.
- Assign digital quizzes as formative assessments.

Extending Learning Beyond the Classroom

Encourage students to observe physical changes in their environment.

Assignments:

- Keep a science journal recording physical changes observed at home or outdoors.
- Research and present on real-world applications of physical changes in industries.
- Design their own experiments related to physical changes and share findings.

Conclusion

Understanding physical changes is a vital component of physical science education. By leveraging creative teaching strategies and resources, such as those provided by Creative Teaching Press, educators can foster curiosity, deepen understanding, and nurture scientific literacy among students. From hands-on experiments to visual aids and real-world connections, teaching physical changes effectively prepares students for more advanced topics and encourages a lifelong appreciation for science.

Remember: The goal is to make science accessible, engaging, and relevant. Through creative teaching approaches, educators can inspire the next generation of scientists and thinkers to explore the fascinating world of physical science.

Question What are some effective creative teaching strategies for explaining physical changes in science classes? Creative strategies include hands-on experiments, visual demonstrations, interactive models, and storytelling to help students grasp concepts like melting, freezing, and condensation in engaging ways. How can teachers make the concept of physical changes more relatable to students? Teachers can use everyday examples such as ice melting, boiling water, or folding paper to illustrate physical changes, making the concepts more familiar and easier to understand. What role does the Physical Science Creative Teaching Press play in improving science education? The Physical Science Creative Teaching Press provides innovative resources, lesson plans, and activities designed to foster creativity and enhance student understanding of physical science concepts like physical changes. What are some common misconceptions students have about physical changes, and how can creative teaching address them? Students often confuse physical changes with chemical changes. Creative teaching methods like experiments and visual aids can clarify that physical changes do not alter the substance's identity, helping to correct misconceptions. Can technology enhance teaching about physical changes in physical science? If so, how? Yes, technology such as virtual labs, simulations, and interactive videos can provide dynamic and visual representations of physical changes, making abstract concepts more accessible and engaging for students.

Related keywords: physical science, creative teaching, physical changes, science education, hands-on experiments, teaching resources, science curriculum, classroom activities, scientific concepts, educational publishing

Read the full article online: [PHYSICAL SCIENCE CREATIVE TEACHING PRESS PHYSICAL CHANGES](#)